

# Formulation Development and Optimization of Fenopropfen Floating Tablet Using QbD Approach

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## ABSTRACT

**Aim:** The study's goal was to develop and characterize a new type of Fenopropfen tablet that can float. **Materials and Methods:** Research on the drug's performance was carried out to evaluate its organoleptic qualities, solubility, melting point, and partition coefficient. Bulk and pharmaceutical formulations of the medication were evaluated using UV spectroscopy to identify Fenopropfen in accordance with ICH Guidelines. Fenopropfen floating tablets were made utilizing the direct compression method with different concentrations of HPMC K100 M, Xanthan gum, and guar gum. Furthermore, a QbD method was used to manufacture the Fenopropfen tablets. To find the optimal formulation of the solid dispersed drug material in Fenopropfen tablets, a 23-full factorial Design of Experiment (DoE) was carried out using Design-Expert 22.0.2.0 software. **Results:** The produced tablets were subjected to a micromeritics research, which included both a pre- and post-compression phase. Using a USP Dissolution test device type II (Paddle), an *in vitro* dissolution research was conducted. There were three months of stability testing done at high temperatures. The research confirmed that Fenopropfen had a distinct odor and a white color look. Its melting point (MP) was 171°C, and it was soluble in DMSO without any difficulty. We measured 0.31gm/cm<sup>3</sup> for the optimized formulation's bulk density, 0.46gm/cm<sup>3</sup> for the tapped density, 14.78 for the Carr index, 1.42 for the hausner ratio, and 30.17 degrees for the angle of repose. The optimized batch (F4) including polymers and excipients floated for a long duration and had a short lag time for buoyancy. There was a 99.12 percent drug release rate after 12 hours in *in vitro* dissolution tests. **Conclusion:** The stability showing that formulation was stable for months of storage and minimal change was found in color, shape, appearance, drug content, *in vitro* dissolution studies and floating lag time value.

**Keywords:** Fenopropfen, Drug design, Polymer, Floating tablet.

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## INTRODUCTION

In order to overcome the drawbacks of conventional drug delivery methods, researchers have developed the Innovative Drug Delivery System (NDDS).<sup>1</sup> To accomplish the required pharmacological effects of a drug in a safe manner, a new technology called Non-traditional Drug Delivery Systems (NDDS) is being developed and implemented.<sup>2</sup> Scientific site-targeting within the body has the potential to increase drug potency, regulate drug release, and produce a more sustained pharmacological effect. Recent therapeutic applications and formulation system development strategies to enhance drug delivery potential were discussed at NDDS.<sup>3</sup>

Due to their excellent efficacy and durability, NDDS have recently garnered much attention, especially in Gastrointestinal (GI)

illnesses.<sup>4</sup> The addition of NDDS into bilayer floating tablets has also increased the use of the FDDS.<sup>5,6</sup> The FDDS approach to drug delivery is simple, inexpensive, and more practical than alternative approaches. The idea is to make the dosage form lighter than the fluids in the stomach so that it floats. Drug Delivery Systems (FDDS) rely heavily on polymers since they are the primary factor in determining stomach retention time and drug protection.<sup>7</sup> The floating system utilizes polymers to direct medicine delivery to a specific area of the gastrointestinal tract, specifically the stomach.

Drugs that have weak solubility and limited stability in intestinal fluids can benefit from floating tablets, which were developed to keep the medicine in the stomach.<sup>8</sup> Nonsteroidal Anti-Inflammatory Drugs (NSAIDs) include fenopropfen. Fenopropfen is prescribed for the treatment of a wide variety of painful conditions, including osteoarthritis, rheumatoid arthritis, and ankylosing spondylitis, as well as for the management of mild to moderate pain.<sup>9</sup> Although it is absorbed fast after oral administration, considerable first-pass metabolism occurs. It



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causes nausea and gastrointestinal discomfort, among other GI adverse effects.<sup>10,11</sup> Its half-life is only 2-3 hr.

Cellulose Ether (CE) is a polymer derived from cellulose that is widely distributed in the natural world. It's a polymer that dissolves in water, too. Gupta *et al.* (2021)<sup>12</sup> and Tudoroiu *et al.* (2021)<sup>13</sup> note that they are also useful as a suspending agent, protective colloids, lubricants, emulsifiers, and surfactants. Hydrophilic Polyethylene Glycol (HPMC) is widely utilized in the pharmaceutical industry as a hydrophilic gel matrix and bio-adhesion material. The fact that it is safe to use, can be compressed easily, and can hold a lot of drugs has contributed to its widespread acceptance.<sup>14,15</sup> It has the properties and potential uses of emulsification, bonding, thickening and adhering, suspension, gelation, and film formation because of its variation in molecular weight and viscosity. When developing products with controlled medication release, HPMC is an excellent polymer to use. Although it remains insoluble at all of the body's pH levels, it expands in the presence of gastric juice. After being exposed to water, the material becomes permeable and can release modified drugs slowly over time.<sup>16,17</sup> This makes it a good option for boosting patients' adherence. This research was conducted to better disperse Fenopropfen into the body's bloodstream and to create longer-acting tablets of the medicine. The drug release was also investigated *in vitro*.

## MATERIALS AND METHODS

### Pre-formulation studies

#### Organoleptic Properties

Human sense organs were used to evaluate organoleptic qualities. Color, smell, visual appeal, and other "organoleptic" factors were all taken into account.<sup>18</sup>

#### Solubility study

Fenopropfen's solubility in various solvents was evaluated qualitatively in accordance with USP NF, 2007. An accurately weighed dose of drug (1 mg) was poured into a 10 mL test tube, and then diluted with the appropriate solvents (1 mL of methanol, ethanol, DMSO, chloroform, and acetone).

#### Melting Point

The open Capillary technique was used to determine the melting point. A little dose of the medication was sealed at one end of a capillary tube that was only 10-15 mm long and around 1 mm in diameter. To ensure that the samples were heated uniformly and gradually, the capillary containing the samples was suspended, and a thermometer was put to monitor the temperature. The melting point is determined by the observed temperature range in which the sample melts.

### Partition coefficient

In n-Octanol:water system, drug partition coefficient was investigated. It was determined by taking 5mg of medication in two separating funnels with 20 mL of n-Octanol and 20 mL water. For balance, the separating funnel was shaken for 2 hr in a wrist action shaker. The drug in the aqueous phase was measured spectrophotometrically at 272 nm. Drug partition coefficient was estimated using this formula:

$$\text{Partition coefficient, } K = \frac{\text{Amount of drug in organic phase}}{\text{Amount of drug in aqueous phase}}$$

### Analytical method development by UV

A process is considered validated if it has been shown, through careful documentation, to reliably and repeatedly yield the intended outcome, with the associated quality attributes and standards, over an extended period of time. Multiple metrics, including linearity, precision, intraday, intraday, ruggedness, and robustness, were used to verify the method's validity.<sup>19</sup>

### Preparation of standard stock solution

Fenopropfen, equivalent to around 5 mg, was weighed and placed in a 5 mL volumetric flask. A solution with a concentration of 1000 g/mL was prepared by diluting the solid in ethanol until the volume reached 5 mL. The standard stock solution concentration is 100 g/mL, which was achieved by taking 1 mL of the stock solution and diluting it to 10 mL with the appropriate solvent.

### Lambda max

A concentration of 20 g/mL was prepared by transferring 2 mL of the aforesaid stock solution into a 10 mL volumetric flask and filling it up to the mark with solvent. UV-vis Spectrophotometer measurements were taken between 200 and 400 nm, with each solvent serving as a blank. Maximum absorbance (max) wavelength was determined.

### Formulation of Fenopropfen floating tablets

Fenopropfen floating tablets were made utilizing the direct compression method with different concentrations of HPMC K100 M, Xanthan gum, and guar gum. Fenopropfen 600 mg per tablet was achieved by adjusting the tablet's weight. A #80 mesh sieve was used to filter all of the particles. The necessary amount of medication and polymers were carefully combined. The final touches were the addition of talc for gliding and 2% magnesium stearate for lubrication. Using a tablet compression machine, the mixture was immediately compressed (punched) to create the tablets. The tablets now measure in at a more manageable 800 mg.<sup>20</sup>

### Optimization of Fenopropfen tablet formulation

By imposing requirements (objectives) on both the response and the factors, an optimal formulation was achieved. To obtain a perfect formula, we used the Box-Behnken Design available

in DESIGN EXPERT 22.0.2.0 (STAT-EASE) edition software. Because of the variable nature of the component mixture, 17 distinct formulation batches were created for testing. Numerous formulations were made and tested for all of the indicated outcomes. Analysis of Variance (ANOVA) was used to compare and contrast the two models to find out which one was more significant. A linear relationship between the response and the factor is indicated by a positive sign before the factor in polynomial equations, while a negative sign indicates the inverse relationship. The correlation between the independent [HPMC K100 M, Xanthan gum, and Guar gum] and dependent [Drug content, Floating lag time] variables was calculated across all 17 experiments.<sup>21</sup>

## Pre-compression Evaluation Parameters

### Bulk Density

A measured amount of powder that has been sieved through a mesh size of 40 and then carefully put into a graduated cylinder. The powder bed was then made uniform by pouring the powder into the graduated cylinder and not stirring the powder. The volume was then read from the cylinder's graduation marks and converted to milliliters. The total measured space was referred to as the bulk volume. The formula was used to determine the density of the bulk material.<sup>22</sup>

$$D_b = M/V_b$$

Where M is the powder's mass and V<sub>b</sub> is its volume in bulk

### Tapped Density

The identical measuring cylinder was used for both the bulk volume and the tap density measurement. The tap density apparatus was run for 500 taps at a rate of 300 taps per minute. After recording the initial volume as (V<sub>a</sub>), 750 taps were recorded as (V<sub>b</sub>). If the discrepancy between V<sub>a</sub> and V<sub>b</sub> is less than 2%, the tapped volume will be calculated based on V<sub>b</sub> is.<sup>22</sup> The formula for tapped density can be found below.

$$D_t = M/V_t$$

Where M is the powder's mass and V<sub>t</sub> is the powder's tapped volume.

### Angle of repose

When calculating the angle of repose, the fixed funnel approach was used. Cones develop when powder is poured through a funnel onto a flat surface. It is called the angle of repose, and it is the angle formed by the sides of the cone and the horizontal. Because it indicates the point at which the interparticle attraction is stronger than the force of gravity on a particle, the angle is used as a proxy for the cohesiveness of the powder. The angle of repose is smaller for a powder that flows freely because the resulting cone will have shallower sides.<sup>22</sup> By plugging in the measured

numbers for the pile's base radius "r" and height "h," we were able to calculate the angle of repose.

$$\Theta = \tan^{-1}(h/r)$$

Here, h = Height of pile,

r = Radius of pile,

Θ = Angle of repose.

### Hausner's Ratio

The measured and mixed powder was weighed, then transferred to the cylinder. After 300 taps, either a constant volume was recorded for the poured bulk volume (V<sub>b</sub>) and the tapped volume (V<sub>t</sub>), or the poured bulk density (P<sub>b</sub>) and the tapped density (P<sub>t</sub>) in g/mL were determined with the aid of a tap density tester. The Hausner ratio was determined by plugging the data for the bulk density and the tapped density into the following formula.

$$\text{Hausner's Ratio} = \text{Tapped density} / \text{Bulk density}$$

### Percentage Compressibility (Carr's Index)

#### Percentage

The percentage of compressibility, expressed as Carr's Index, was determined by multiplying the ratio of the tapped density to the bulk density by 100. How to Determine a Carr's Index (CI).

$$CI (\%) = \left[ \frac{\text{Tapped density} - \text{Bulk density}}{\text{Tapped density}} \right] \times 100$$

## Post-compression Evaluation

### General appearance

The color, shape, and overall look of the tablets are assessed once they have been formulated.<sup>22</sup>

### Thickness

To measure the tablets' thickness and diameter, a vernier caliper was employed. We utilized an average of 5 pills from each formulation type to get our results. The unit of measure is the mm.

### Hardness

The "force required to break a tablet in the diametric compression test" is the standard measure of hardness. The ability to crush a tablet is synonymous with hardness. Tablets' hardness determines how well they hold up throughout storage, shipping, and handling before to use. The Monsanto tablet hardness tester was used to get accurate results. The tablet was placed between two anvils of a hardness tester (Monsanto), and the force applied to them was increased incrementally until the tablet broke. The pressure at which the tablet breaks was measured using the clearly labeled

scale. Kg/cm<sup>2</sup> was the unit of measure for hardness. Lists the results of a hardness test.<sup>22</sup>

### Weight variation

According to the protocol, 20 pills were weighed at random on an electric balance, and their average weight was then compared to the weights of the individual tablets.

### Friability test

The capacity of tablets to survive abrasion during packaging, handling, and transit is measured by this test. Insufficiently cohesive tablet components are usually the cause of friability. The Roche Friabilator is a circular plastic chamber separated into 2-3 compartments, and the first step is to weigh and place 10 tablets within. Tablets are dropped from a height of 1.5 meters (4.9 feet) and the capsule rotates 100 times in 4 min. After then, the tablets undergo a second round of weighing.<sup>22</sup> The percentage represents the difference in weight that was measured. It ought to be 1.0 or below. Table 17 reports an analysis of friability.

$$\% \text{ Friability} = \frac{W1 - W2}{W1} \times 100$$

Where,

W1= weight of tablets before test,

W2 = weight of tablets after test.

### Drug content

UV-visible spectrophotometric analysis is used to determine the total amount of medication present in the formulation. Phosphate buffer pH 7.4 was used to dissolve the formulation, which was then agitated for 5 min before enough of the buffer was added to bring the total amount to 100 mL. After 15 min, the solution was filtered using Whatman filter paper after being sonicated. We made dilutions of the solution and evaluated its absorbance spectrophotometrically at 272 nm against 0.01N HCl as a blank using a UV/visible spectrophotometer.

### In vitro Buoyancy Test

The produced tablets were tested for buoyancy in a container of 200 mL 0.1 N HCl (pH 1.2, temp. 37.0°C) in a 250 mL beaker. Visual observation was used to determine the lag time and total buoyancy time from when the dosage form was introduced and when it became buoyant in the medium, as well as the floating durations of tablets. Total Floating Time (TFT) is the amount of time that elapses after a dosage form first floats to the surface of a medium, while Floating Lag Time (FLT) is also known as Buoyancy Lag Time (BLT).<sup>23</sup>

### In vitro dissolution studies

The dissolution tests were performed in a phosphate buffer at pH 7.4 in a volume of 900 mL using a USP type II apparatus (USP

XXIII Dissolution Test Apparatus). The dissolution medium was kept at 37±2°C throughout the experiment. At predefined intervals, 2 mL aliquots of sample were removed from the dissolution equipment and replaced with new dissolution medium. Following appropriate dilution in phosphate buffer pH 7.4, absorbance was measured at 272 nm using a UV/visible Spectrophotometer on the samples that were collected. The standard calibration curve was used to determine the percentage of medication release.

### Stability studies

The tablet formulations were packed and placed in the stability test chamber for three months of accelerated stability testing at (30°C±2°C and 60 ± 5% RH) and (40°C±2°C and 70 ± 5% RH). The tablets were tested for organoleptic features as color, shape, appearance, drug content, *in vitro* dissolution studies, and floating lag time at 30, 45, 60, and 90 days (3 months). In compliance with International Conference on Harmonization (ICH) criteria, optimal formulation tablets were tested for stability under accelerated storage conditions for 3 months. The optimal formulation of 0 days was used to compare all results.

## RESULTS AND DISCUSSION

### Preformulation studies

#### Organoleptic properties

An evaluation of the API's organoleptic qualities, including color, odor, appearance and state, was conducted. Fenopfen was found white in color and had a characteristic odor. Fenopfen exhibited the same color, appearance, and odor as the I.P. requirements for these characteristics. Preformulation studies of fenopfen were in agreement with the findings reported by Hirsch *et al.*, 1978.<sup>24</sup>

#### Solubility study

Fenopfen's solubility was tested in a wide range of volatile and nonvolatile liquids, including methanol, ethanol, chloroform, and acetone. Based on the findings, it can be concluded that the medication is soluble in ethanol and Dimethyl sulfoxide but insoluble in water.

#### Melting point study

The melting point of a substance can be calculated using the capillary method. Fenopfen's melting point was measured to be 171°C, which is within the acceptable range for this medication.

#### Determination of Partition coefficient

The drug's partition coefficient in n-Octanol:water was studied. It is the proportion of the unionized drug that exists in the aqueous phase relative to the organic phase at equilibrium. The drug's lipophilic and hydrophilic properties can be described using the partition coefficient. Lipophilic drugs have P values significantly higher than 1, while hydrophilic drugs have P values significantly

lower than 1. Fenopfen was reported to have partition coefficients of 3.47.

### UV-spectroscopy analysis of Fenopfen

UV-visible spectrophotometer (Shimadzu-1700) was used to determine wavelength (absorption maxima) of a substance. The wavelength of the Fenopfen was found to be 272 nm (Figure 1).

### Calibration curve

Least-squares linear regression analysis of the calibration curve verified the linearity of the suggested approach. By graphing absorbance against Fenopfen concentration from 10-70 g/mL, we were able to derive the regression equation for Fenopfen. In a concentration range of 10-70 ng/mL for the drug, a seven-point calibration curve was obtained. In the concentration range under study, the drug's reaction was determined to be linear, with a linear regression equation of  $y = 0.005x + 0.050$  and a correlation coefficient of  $R^2 = 0.991$  (Figure 2).

### Preparation and optimization of Fenopfen floating tablets

Table 1 showing different batches of Fenopfen floating tablets. The drug quantity was 600 mg in all batches. The quantity of HPMC K100 M, Xanthan gum and guar gum were vary between 30-50 mg, 20-40 mg and 20-40 mg, respectively. Formulation trails as per Box- Behnken design is shown in Table 5. The response of drug content and floating lag time was recorded as per three variables. The highest drug content was achieved for the formulation F13 i.e., 99.81%, however, floating lag time was highest for F12 i.e., 37.04 sec. The relationship between the independent and dependent variables, as well as all the results from the 17 runs.

### Pre-compression study of Fenopfen

The regularity of the tablet's weight could be affected by the powder's flow, therefore it's an important factor to consider. Therefore, the number of polymers needed to keep acceptable flow and compaction properties was calculated by measuring the flow properties of the powder blend of the optimized tablet formulation. The angle of repose, bulk density, tapped density, Carr's index, and Hausner's ratio were among the pre-compression parameters tested on the powder mix of the formulation. For a pre-compression investigation, 17 distinct Fenopfen formulations were analyzed by measuring their bulk density, tapped density, Hausner's ratio, Carr's ratio, and angle of repose, among other parameters. F12 has the largest bulk density, with  $0.39 \text{ g/cm}^3$ . The highest values for tapped density ( $0.37 \text{ g/cm}^3$ ), Hausner's ratio ( $0.41 \text{ g/cm}^3$ ), Carr's ratio (16.08%), and angle of repose ( $31.47^\circ$ ) were all found in F14 ( $0.37 \text{ g/cm}^3$ ), F13 ( $0.41 \text{ g/cm}^3$ ), F8, and F2, respectively. Tapped density, floating lag

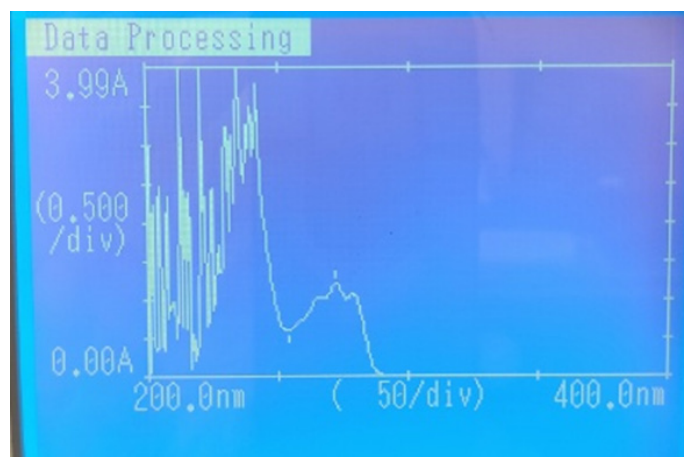


Figure 1: UV-spectroscopy of Fenopfen.

Table 1: Composition of different variables used for formulation.

Ingredients	F1	F2	F3	F4	F5	F6	F7	F8	F9	F10	F11	F12	F13	F14	F15	F16	F17
Drug	600	600	600	600	600	600	600	600	600	600	600	600	600	600	600	600	600
HPMC K100 M	30	30	30	40	40	40	40	30	40	40	40	50	50	50	40	50	40
Xanthan gum	30	40	30	30	20	30	30	20	30	40	20	20	30	30	40	40	30
guar gum	40	30	20	30	40	30	30	30	30	20	20	30	40	20	40	30	30
Sod. bicarbonate	30	30	30	30	30	30	30	30	30	30	30	30	30	30	30	30	30
Citric acid	15	15	15	15	15	15	15	15	15	15	15	15	15	15	15	15	15
Magnesium Stearate (2%)	16	16	16	16	16	16	16	16	16	16	16	16	16	16	16	16	16
Talc (2%)	16	16	16	16	16	16	16	16	16	16	16	16	16	16	16	16	16
Lactose	q.s	q.s	q.s	q.s	q.s	q.s	q.s	q.s	q.s	q.s	q.s	q.s	q.s	q.s	q.s	q.s	q.s
Total weight	800	800	800	800	800	800	800	800	800	800	800	800	800	800	800	800	800

time, Carr's index, and Hausner's ratio are some of the metrics whose results are displayed in Table 2.

### Post-compression evaluation of tablets

Post-compression factors including tablet appearance, weight fluctuation, hardness, thickness, friability, and drug content were compared between formulations. It was determined that the thickness of several formulations ranged from 3.2 mm to 4.4 mm. Hardness between range 3.3 - 4.3 kg/cm<sup>2</sup>, Friability between 0.68-0.77%, Weight variation between 797.41-804.59 mg. The drug content was found to be between 94.33-99.58% (Table 3). The flow properties of Fenopropfen tablets were determined to be within the range indicative of better flow property based on post-compression metrics such as bulk density, tapped density, Carr's index, and Hausner's ratio.

### In vitro buoyancy and floating test

The pills did not dissolve when submerged in a 0.1N HCl solution at 37°C with a pH of 1.2. Buoyancy Lag Time (BLT) and Total Floating Time (TFT) were both found to be satisfactory for the mixture of polymers and excipients. Figure 3 displays that the buoyancy lag times for formulations F1 through F17 ranged from 24.65 sec to 37.04 sec. It was highest for F12 i.e., 37.04 sec. Floating time was ranged between 10.05-12.42 hr. F14 showed floating time of 12.42 hr (Figure 4).

### In vitro dissolution studies

*In vitro* dissolution study of prepared tablets is shown in Table 4. The dissolution of tablets was observed for 12 hr. It was found that nearly all the tablet formulations were ~99% dissolved upto 12 hr.

### Optimization of Fenopropfen tablet formulation

Constraints (goals) were placed on both the dependent (response) and independent (factors) variables to arrive at the optimum formulation. Table displays the limitations on the possible answers and contributing factors. Using DESIGN EXPERT 22.0.2.0 (STAT-EASE), we were able to narrow down the optimal formulation options to a single program. Due to the experimental character of this component mixture, 17 distinct formulation batches have been produced (when counting construction centers). Numerous formulation lots were made, as indicated, and evaluated based on the given replies. Analysis of Variance (ANOVA) is used to compare the model to another model and determine its significance. A positive sign before a factor in a polynomial equation indicates a linear relationship between the response and the factor, while a negative sign indicates the opposite. The relationship between independent and dependent variables was analyzed, along with all the results from 17 separate runs.

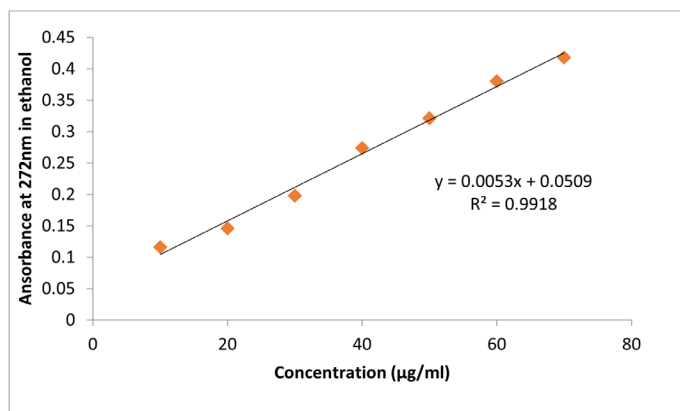


Figure 2: Calibration curve of Fenopropfen.

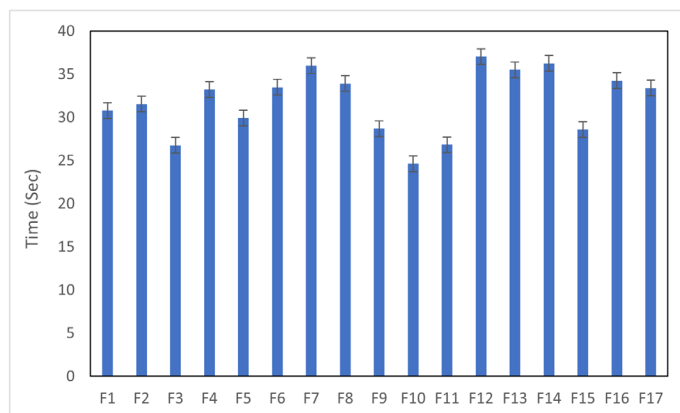


Figure 3: *In vitro* buoyancy time of prepared tablets.

### Effect of formulation variables on Drug content

When the design formulation findings were fitted into many models, a linear model was significant for drug content ( $F = 3.45$ ,  $P = 0.0484$ ). In this model, the drug content was significantly changed by factors A and B (polymers), but not by component C (the amount of Guar gum). This is the model equation: Table 6 shows the formula for calculating the number of drugs in a sample: drug content = +97.46 -0.5500 A + 1.15 B -0.1650 C. Figure 5 depicts a three-dimensional response surface map that helps to explain the impact of components A and B.

### Sum of squares is Type III - Partial

A Model F-value of 3.45 indicates statistical significance. An F-value this high could only arise from random chance 4.84% of the time.

Model terms are statistically significant when their p-values are less than 0.0500. Here, the model term B is particularly important. Model terms are not significant if their p-values are bigger than 0.1000. If your model has a large number of meaningless terms (except those necessary to support hierarchy), you may benefit from performing a model reduction.

A Lack of Fit F-value of 5.35 indicates a 6.06% possibility that such a significant number could be attributed to random chance

**Table 2: Pre-compression evaluation of Fenoprofen.**

Formulation	Bulk density (g/cm <sup>3</sup> )	Tapped density (g/cm <sup>3</sup> )	Hausner's ratio	Carr's ratio%	Angle of repose
F1	0.34	0.38	1.17	13.03	27.35
F2	0.32	0.46	1.16	13.01	31.47
F3	0.27	0.42	1.16	12.05	28.72
F4	0.31	0.42	1.14	14.08	29.85
F5	0.32	0.43	1.15	11.01	30.15
F6	0.29	0.47	1.16	12.04	30.18
F7	0.34	0.46	1.17	15.05	29.77
F8	0.34	0.41	1.14	16.08	30.07
F9	0.32	0.39	1.13	14.07	29.86
F10	0.35	0.41	1.17	13.08	26.97
F11	0.38	0.37	1.14	13.06	30.58
F12	0.39	0.41	1.14	14.12	29.72
F13	0.32	0.41	1.18	12.11	30.76
F14	0.37	0.47	1.16	13.01	29.42
F15	0.28	0.46	1.16	12.08	25.46
F16	0.27	0.46	1.12	14.18	30.56
F17	0.29	0.44	1.14	12.17	26.39

**Table 3: Post-compression evaluation of tablet.**

Formulation	Thickness (mm)	Hardness kg/cm <sup>2</sup>	Friability (%)	Weight variation (mg)	Drug Content (%)
F1	3.2	4.1	0.75	797.41	96.46
F2	3.8	4.2	0.76	800.15	98.51
F3	3.3	3.9	0.75	803.24	96.71
F4	4.1	3.8	0.76	801.54	97.78
F5	3.5	3.8	0.72	803.15	95.92
F6	3.4	3.8	0.68	801.61	98.78
F7	3.4	4.0	0.76	800.77	97.45
F8	4.2	4.1	0.73	804.59	98.56
F9	3.6	3.5	0.72	800.57	98.01
F10	4.3	3.3	0.69	798.85	99.58
F11	4.2	3.6	0.68	801.64	97.38
F12	3.9	3.4	0.71	797.54	94.33
F13	4.1	4.3	0.73	802.06	97.81
F14	3.3	3.3	0.77	801.34	96.25
F15	3.3	3.6	0.74	798.24	99.13
F16	3.3	3.5	0.73	800.57	97.45
F17	4.4	4.2	0.74	803.76	97.38

Table 4: *In vitro* dissolution studies of fenopropfen tablets.

Time (hrs)	F1	F2	F3	F4	F5	F6	F7	F8	F9	F10	F11	F12	F13	F14	F15	F16	F17
30 min	12.23	18.29	19.52	23.28	19.23	19.95	19.6	19.38	15.39	18.29	18.29	19.23	16.22	18.29	16.24	12.17	18.29
1	28.36	26.19	27.42	48.91	27.13	27.85	27.5	27.28	31.52	26.19	26.19	38.13	32.35	26.19	32.37	27.21	26.19
2	35.61	33.72	34.95	61.11	34.66	35.38	35.03	34.81	38.77	33.72	33.72	44.72	39.6	33.72	39.62	30.94	33.72
3	41.84	41.21	42.44	33.14	42.15	42.87	42.52	42.3	45	41.21	41.21	53.21	45.83	41.21	45.85	31.96	41.21
4	46.25	53.54	54.77	51.82	54.48	55.2	54.85	54.63	49.41	53.54	53.54	67.14	50.24	53.54	50.26	40.96	53.54
5	52.36	61.19	62.42	67.62	62.13	62.85	62.5	62.28	55.52	61.19	61.19	71.62	56.35	61.19	56.37	46.88	61.19
6	61.84	74.34	75.57	79.45	75.28	76.02	75.65	75.43	65.26	74.34	74.34	76.54	65.83	74.34	65.85	56.23	74.34
7	69.22	83.24	84.47	85.63	84.18	84.9	84.55	84.33	72.38	83.24	83.24	86.67	73.21	83.24	73.23	60.53	83.24
8	79.27	91.29	92.52	89.06	92.23	92.95	92.6	92.38	82.43	91.29	91.29	89.16	83.26	91.29	83.28	68.94	91.29
9	83.51	96.27	97.5	93.86	97.21	96.93	97.58	97.36	86.67	96.27	96.27	93.46	87.5	94.27	87.52	76.83	93.27
10	87.21	98.03	99.2	96.62	97.75	97.34	96.38	99.07	90.37	96.91	98.81	96.53	99.2	96.67	91.22	88.19	96.89
11	97.88	99.78	99.6	99.06	98.79	98.07	98.86	99.12	98.37	99.25	99.83	99.06	99.6	97.26	99.89	9686	98.37
12	99.05	99.86	99.87	99.22	99.02	99.57	98.98	99.32	99.11	99.9	99.98	99.23	99.8	99.76	99.97	99.01	99.09

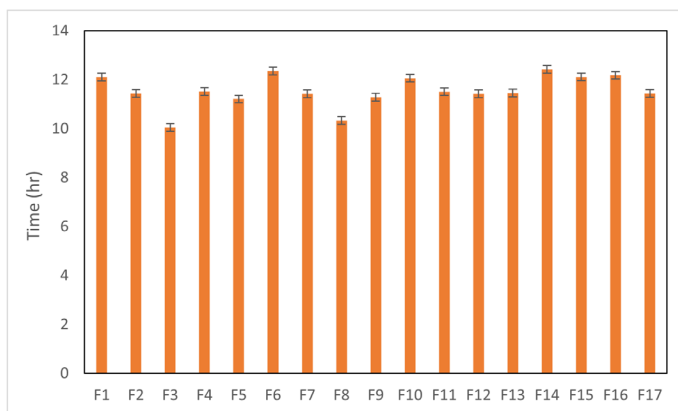


Figure 4: *In vitro* floating time of prepared tablets.

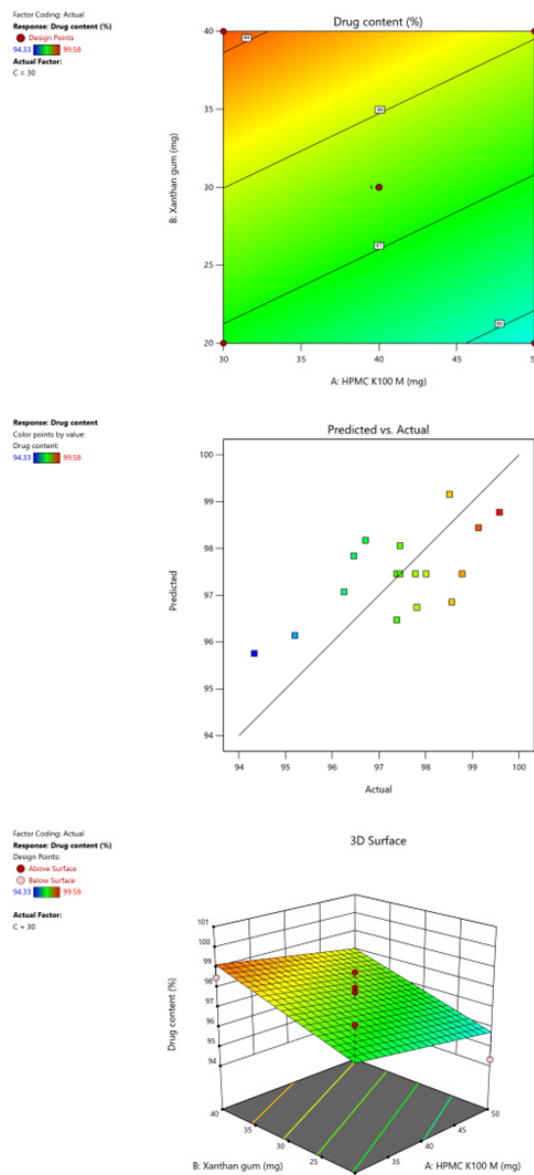


Figure 5: Response surface plot showing combined effect of xanthan gum and HPMC K100 M on drug content of formulation.

alone. Inadequate model fit is undesirable. This low likelihood (below 10%) raises some red flags.



### Effect of formulation variables on Floating lag time

When the design's specified formulations were entered into a number of models, it was discovered that the quadratic model was significantly better in predicting the Floating lag time, with a *F* value of 3.68 and a *P* value of 0.0500. This is the model equation: Time delay while floating = +32.97 + 2.51 A - 1.08 B + 1.29 C - 0.0975 AB - 1.19 AC + 0.2150 BC + 3.03 A<sup>2</sup> - 1.81 B<sup>2</sup> - 3.66 C<sup>2</sup>. The analysis of variance data is presented in Table 7. Numerically optimized solutions were produced from the analysis and optimization of these experimental results. A solution was chosen at random from the numerical optimization results, labeled as optimized formulation, and taken into account (Figure 6). Predicted and experimental findings are presented, demonstrating how closely they align. It was found that the

observed reaction was quite near to the response predicted by the software.

### Sum of squares is Type III - Partial

There is statistical significance in the model, since the *F*-value is 3.68. A big *F*-value is extremely unlikely to be the result of random chance; the probability is only 5.00%.

Significance of model terms is indicated by *p*-values less than 0.0500. A, A<sup>2</sup>, and C<sup>2</sup> all play important roles as model terms here. The model terms are not significant if their *p*-values are bigger than 0.1000. Model reduction is useful if your model contains a large number of meaningless terms (excluding those necessary to maintain hierarchy).

**Table 5: Formulation trials as per Box-Behnken design.**

Run	Factor 1 A: HPMC K100 M (mg)	Factor 2 B: Xanthan gum (mg)	Factor 3 C: Guar gum (mg)	Response 1 Drug content	Response 2 Floating lag time (Sec.)
1	30	30	40	96.46	30.79
2	30	40	30	98.51	31.54
3	30	30	20	96.71	26.78
4	40	30	30	97.78	33.23
5	40	20	40	95.2	29.93
6	40	30	30	98.78	33.5
7	40	30	30	97.45	36.01
8	30	20	30	98.56	33.92
9	40	30	30	98.01	28.71
10	40	40	20	99.58	24.65
11	40	20	20	97.38	26.83
12	50	20	30	94.33	37.04
13	50	30	40	97.81	35.52
14	50	30	20	96.25	36.27
15	40	40	40	99.13	28.61
16	50	40	30	97.45	34.27
17	40	30	30	97.38	33.41

**Table 6: Response 1: Drug content (ANOVA for Linear model).**

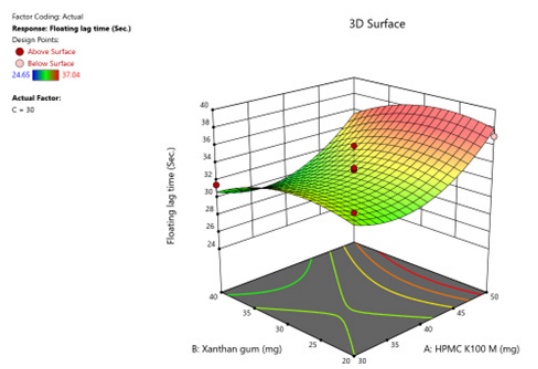
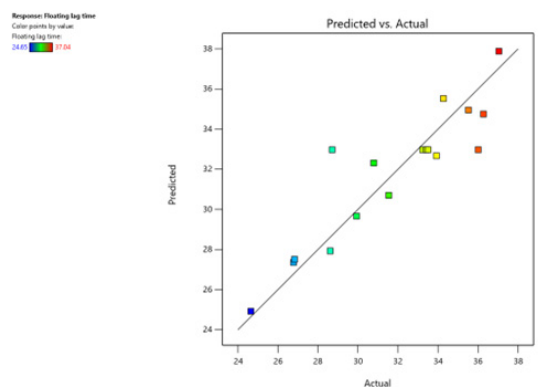
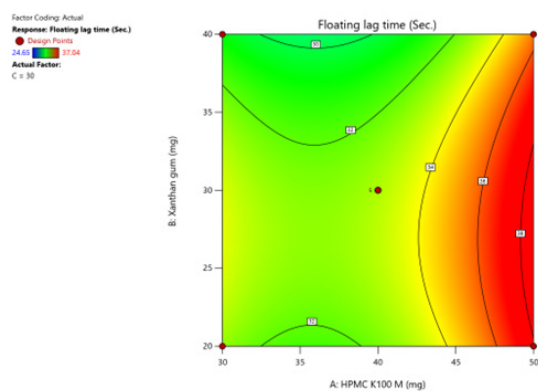
Source	Sum of Squares	<i>d<sub>f</sub></i>	Mean Square	<i>F</i> -value	<i>p</i> -value	
Model	13.22	3	4.41	3.45	0.0484	Significant
A-HPMC K100 M	2.42	1	2.42	1.90	0.1918	
B-Xanthan gum	10.58	1	10.58	8.29	0.0129	
C-Guar gum	0.2178	1	0.2178	0.1707	0.6863	
Residual	16.59	13	1.28			
Lack of Fit	15.32	9	1.70	5.35	0.0606	Not significant
Pure Error	1.27	4	0.3180			
Cor Total	29.81	16				

With an *F*-value of 0.52 for Lack of Fit, this is not statistically different from the pure error. An *F*-value for Lack of Fit this high has a 69.08% likelihood of being due to random chance. An insignificant mismatch between the data and the model is ideal.

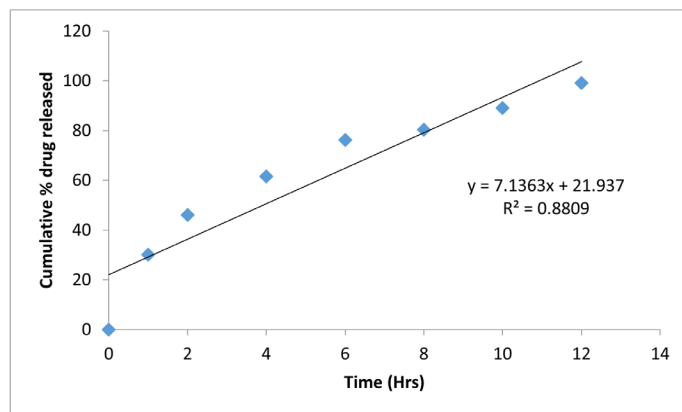
### Evaluation parameter of optimized formulation

After compression, the optimal formulation was tested for quality indicators like tablet color, shape retention, hardness, thickness, friability, disintegration time, and medication content. The capsule-shaped, white pills were a product of careful formulation. The design of each tablet was sleek and sophisticated. The size and shape of the tablets were standardized according to the formulation's thickness once it had been optimized. The vernier caliper measurements for the entire batch of tablets showed a consistent range of 3.6 mm. The findings suggested that the

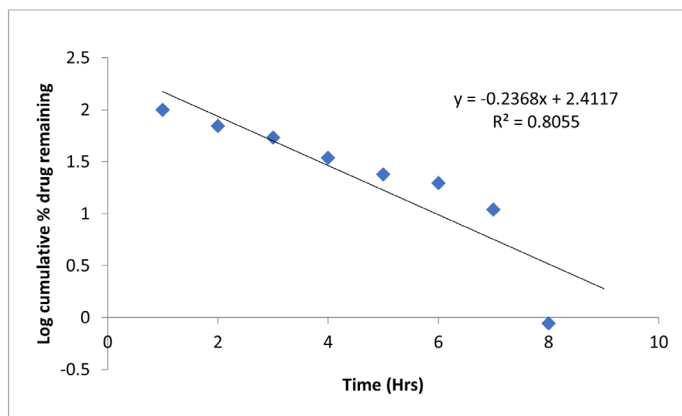
shape and size of the formulation was consistent throughout. The formulation's hardness of 4.4 kg/cm<sup>2</sup> demonstrates the tablets' robust mechanical properties. A total of 800 mg was used in the finished pill. Tablets from this formulation have weight discrepancies that are well within IP's tolerance levels. Every tablet formulation was found to have a friability percentage of 0.74%. The results showed that the mechanical resistance of tablets was high across the board, with no formulation failing to meet the standard by a margin of more than 1%. With a medication content percentage of 98.54%, the formulation is



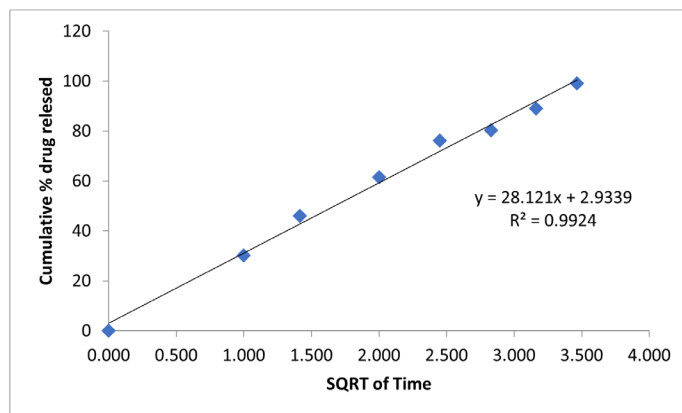
**Figure 6:** Response surface plot showing combined effect of HPMC K 100 M and xanthan gum on Floating lag time of formulation.



**Figure 7:** Zero order kinetic model for fenoprofen tablets.



**Figure 8:** First Order kinetic model for fenoprofen tablets.



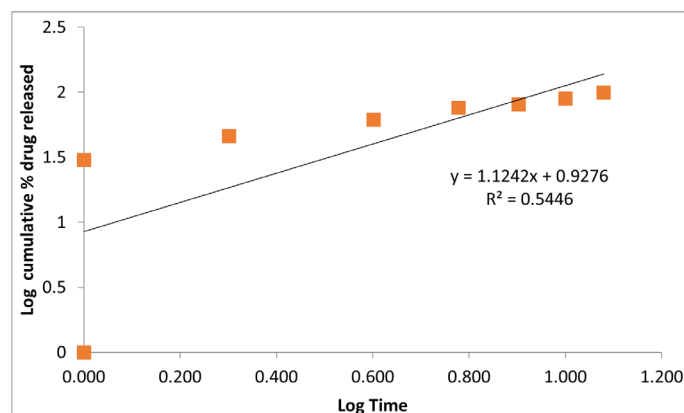
**Figure 9:** Higuchi model for fenoprofen tablets.

guaranteed to have a constant dose of the active ingredient. Based on the outcomes, it seems the formulation is acceptable as per IP standards. When tablets were immersed in a 0.1N HCl solution at a pH (1.2) temperature of 37°C, they floated and stayed afloat without dissolving, indicating that their formulation was optimized. Buoyancy Lag Time (BLT) and Total Floating Time (TFT) were both satisfactory in the optimized batch containing polymers and excipients.

### **In vitro drug (dissolution) kinetics study of optimized formulation**

*In vitro* drug release from the best formulation is shown in Table 8. There was a 99.12% medication release up to 12 hr. At 12 hr, the drug's log cumulative release percentage was 1.996%. The following graphs were generated for the kinetic analysis: cumulative percent drug release versus time (zero order kinetic models); log cumulative percent drug remaining versus time (first order kinetic models); cumulative percent drug release versus square root of time (Higuchi model); log cumulative percent

drug release versus log time (Korsmeyer-Peppas model). Figure 7 through Figure 10 displays all plots. Constant drug release from a drug delivery system, regardless of concentration, is described using zero-order kinetic models. The improved formulation's zero-order graph demonstrated a constant drug release from the



**Figure 10:** Korsmeyer peppas for fenoprofen tablets.

**Table 7: ANOVA for Quadratic model.**

Source	Sum of Squares	$d_f$	Mean Square	F-value	p-value	
Model	183.61	9	20.40	3.68	0.0500	Significant
A-HPMC K100 M	50.35	1	50.35	9.08	0.0196	
B-Xanthan gum	9.35	1	9.35	1.69	0.2353	
C-Guar gum	13.31	1	13.31	2.40	0.1653	
AB	0.0380	1	0.0380	0.0069	0.9363	
AC	5.66	1	5.66	1.02	0.3459	
BC	0.1849	1	0.1849	0.0333	0.8603	
A <sup>2</sup>	38.60	1	38.60	6.96	0.0335	
B <sup>2</sup>	13.75	1	13.75	2.48	0.1594	
C <sup>2</sup>	56.39	1	56.39	10.17	0.0153	
Residual	38.83	7	5.55			
Lack of Fit	10.90	3	3.63	0.5204	0.6908	Not significant
Pure Error	27.93	4	6.98			
Cor Total	222.45	16				

**Table 8: Release kinetics study of optimized formulation.**

Sl. No.	Time (Min.)	cumulative % drug released	% drug remaining	Square root time	log Cumulative % drug remaining	log Cumulative % drug released	log time
1.	0	0	100	0.000	2.000	0.000	0.000
2.	1	30.12	69.88	1.000	1.844	1.479	0.000
3.	2	46.01	53.99	1.414	1.732	1.663	0.301
4.	4	61.56	38.44	2.000	1.585	1.789	0.602
5.	6	76.16	23.84	2.449	1.377	1.882	0.778
6.	8	80.31	19.69	2.828	1.294	1.905	0.903
7.	10	89.08	10.92	3.162	1.038	1.950	1.000
8.	12	99.12	0.88	3.464	-0.056	1.996	1.079

**Table 9: Stability Study of optimized tablet formulation.**

Sl. No.	Time (Days)	30°C±2°C and 60 ± 5% RH			40°C±2°C and 70 ±5% RH		
		Drug content (%)	Floating Lag Time (Sec.)	Drug release (%) 12 hr	Drug content (%)	Floating Lag Time (Sec.)	Drug release (%) 12 hr
1.	0	99.02	30.27	99.12	99.02	30.27	99.12
2.	30	98.94	30.25	99.15	98.04	30.26	99.11
3.	45	99.12	30.27	99.01	98.95	30.27	99.08
3.	60	99.00	30.18	98.97	98.98	30.18	99.97
4.	90	99.14	30.20	99.16	99.14	30.21	99.07

tablet; the zero-order model's  $R^2$  value was 0.880. When the rate of release depends on the concentration of the system, a first order kinetic model is used to characterize the release. The first order kinetic model yielded the following results:  $y = -0.236x + 2.411$   $R^2 = 0.805$ . The constraints of drug delivery and transportation are modeled after the Higuchi equation. The formula for the Higuchi model was determined to be  $y = 28.12x + 2.933R^2 = 0.992$ . Korsmeyer and Peppas's kinetic model produced the following results:  $y = 1.124x + 0.927$   $R^2 = 0.544$ . Drug dissolution or release experiments were performed *in vitro* utilizing specialized equipment. Phosphate buffer, pH 7.4, was used for the *in vitro* drug dissolution testing of the improved formulation. The projected findings were confirmed by the improved formulation's 99.12% drug release within 12 hr. The  $R^2$  value may be seen in row 36 of the table above. The Higuchi kinetic model was found to have the best match, as measured by the coefficient

### Stability study

Optimized Formulation was stability tested for three months at 30°C±2°C and 60% RH and 40°C±2°C and 70 ±5% RH, per ICH recommendations. At accelerated stability conditions, tablets with the optimized formulation were physically and chemically stable for 3 months. Table 9 shows test and other stability study results at various time periods.

### CONCLUSION

The direct compression method was used to make fenopfen floating tablets with HPLC and xanthan gum as polymers at optimum concentrations. The present study found that tablets containing Fenopfen alone are unable to effectively control its release for 12 hr, but that this problem is solved by combining the drug with polymers, which helped in increasing the viscosity of the formulation in the dissolution medium. In terms of how well each formulation-controlled drug release, it was determined that the F4 formulation, which included 40% HPMC, was the most effective. According to the results, formulation F4 demonstrates the ideal swelling properties and drug release kinetics, particularly in the form of zero-order release and floating behavior.

### CONFLICT OF INTEREST

The authors declare that there is no conflict of interest.

### ACKNOWLEDGEMENT

We are thankful to the institute for providing all facilities required during conducting this experiment.

### ABBREVIATIONS

**QbD:** Quality by design; **DoE:** Design of Experiment; **MP:** Melting point; **NDDS:** Novel drug delivery system; **GI:** Gastrointestinal; **NSAIDs:** Nonsteroidal anti-inflammatory drugs; **CE:** Cellulose ether; **HPMC:** Hydroxy Propyl Methyl Cellulose; **ANOVA:** Analysis of variance; **ICH:** International Conference on Harmonization; **BLT:** Buoyancy lag time; **TFT:** Total floating time.

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